



Structures and Bonding of Lanthanide-Doped Endohedral Borospherenes $\text{Ln}@\text{B}_{40}^{0/+}$ ($\text{Ln} = \text{Ce}, \text{Pr}, \text{Nd}, \text{Pm}, \text{Sm}, \text{Eu}, \text{Gd}$)

Xiao-Ni Zhao¹ · Ting Zhang¹ · Xiao-Qin Lu² · Si-Dian Li¹

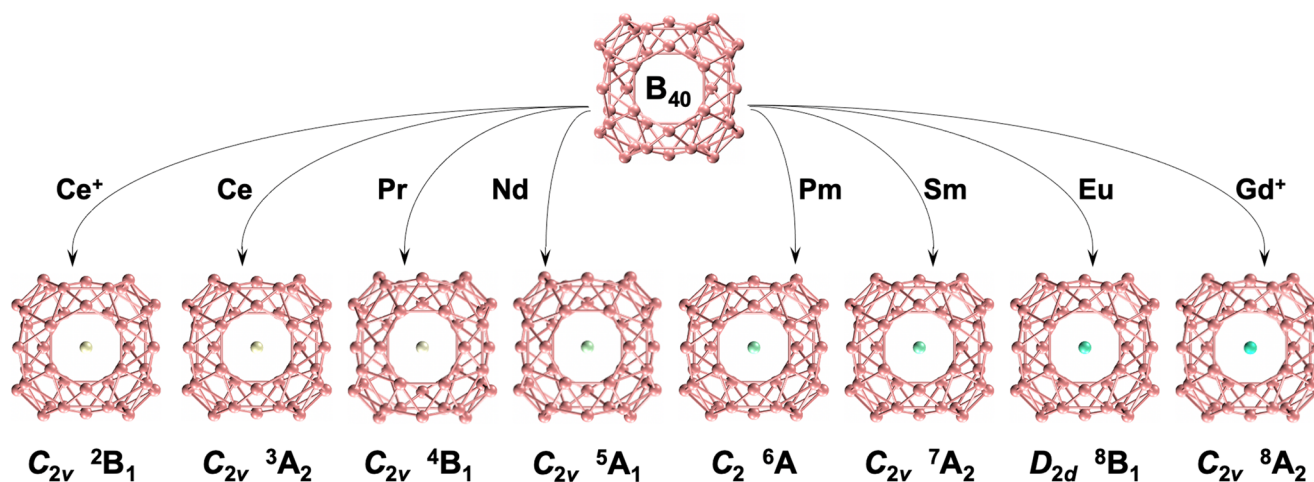
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Abstract

Metal-doped endohedral borospherenes $\text{M}@\text{B}_{40}$ have attracted considerable attention since the discovery of the first borospherenes $\text{B}_{40}^{-/0}$ in 2014. Systematical density functional theory investigations performed herein unveil the ground-state structures and coordination bonding patterns of a series of lanthanide-doped endohedral borospherenes $\text{Ln}@\text{B}_{40}^{0/+}$, including the doublet C_{2v} $\text{Ce}@\text{B}_{40}^{+}$ (**1**, $^2\text{B}_1$), triplet C_{2v} $\text{Ce}@\text{B}_{40}$ (**2**, $^3\text{A}_2$), quartet C_{2v} $\text{Pr}@\text{B}_{40}$ (**3**, $^4\text{B}_1$), quintet C_{2v} $\text{Nd}@\text{B}_{40}$ (**4**, $^5\text{A}_1$), sextet C_2 $\text{Pm}@\text{B}_{40}$ (**5**, ^6A), septet C_{2v} $\text{Sm}@\text{B}_{40}$ (**6**, $^7\text{A}_2$), octet D_{2d} $\text{Eu}@\text{B}_{40}$ (**7**, $^8\text{B}_1$), and octet C_{2v} $\text{Gd}@\text{B}_{40}^{+}$ (**8**, $^8\text{A}_2$). Detailed principal interaction spin orbital (PISO) and adaptive natural density partitioning (AdNDP) bonding pattern analyses indicate that, with the number of unpaired α -electrons changing from $n_\alpha = 1, 2, 3, 4, 5, 6, 7$ to 7 in the series, their coordination bonding energies decrease monotonically from $E_c = 7.22, 6.93, 5.67, 4.85, 4.67, 4.29, 4.02$, to 2.07 eV, respectively, with the dominating percentage contributions of the Ln 5d-involved PISOs to the overall E_c increasing almost monotonically from 66 to 83%, while the minor contributions of the Ln 4f-involved PISOs varying between 0.3% and 12.1% and that of Ln 6s-involved PISO pairs remaining basically unchanged in a narrow range between 6% and 8%. In average, the dominating 5d-involved PISOs in $\text{Ln}@\text{B}_{40}$ contribute about 72.8% to the overall E_c , 19.3% higher than that (53.4%) of the 6d-involved PISOs in the newly reported actinide-doped $\text{An}@\text{B}_{40}^{+/0/-}$, while the minor 4f-involved PISOs in $\text{Ln}@\text{B}_{40}$ contribute about 6.1% to E_c , 15.7% lower than that (21.8%) of the 5f-involved PISOs in $\text{An}@\text{B}_{40}^{+/0/-}$, quantitatively unveiling the differences in coordination bonding patterns between $\text{Ln}@\text{B}_{40}^{+/0}$ and $\text{An}@\text{B}_{40}^{+/0/-}$.

Graphical Abstract



Keywords Lanthanides · Metallo-Borospherenes · First-Principles theory · Structures · Coordination bonding patterns

Extended author information available on the last page of the article

Introduction

The discovery of the first all-boron fullerenes D2d B40-/0 in 2014 [1] and C3/C2 B39- in 2015 [2] marks the onset of borospherene chemistry. In the past one decade, the study of metal-doped B40 has involved integrating main group and transition metals into the borospherene cages in both exohedral and endohedral coordination patterns [3–16] including the newly reported actinide-doped endohedral metallo-borospherenes An@B400/+/- (An = U, Np, Pu, Am, Cm) [12] investigated through detailed bonding pattern analyses. Among these advancements, lanthanides (Ln)-doped endohedral borospherenes Ln@B400/+/- have emerged as a particularly interesting area, driven by the partially occupied 4f electronic configurations of lanthanide elements that render distinctive magnetic and catalytic properties to the systems. Doublet endohedral rare-earth-metal-doped Cs Sc@B40, C2v Y@B40, and C2v La@B40 were proposed at density-functional theory (DFT) level in 2015 [16]. Endohedral octet Eu-doped D2d Eu@B40 (8B2) and septet Gd-doped Cs Gd@B40 (7A") have also been predicted in DFT theory [14]. Joint experimental and theoretical investigation indicated that the perfect spherical trihedral metallo-borospherenes D3h Ln3B18- (Ln = La, Tb) feature three lanthanide centers coordinated in three adjacent equivalent η^{10} -B10 decagons on the cage surface [17]. Core-shell-like La3@[B2@B17]- and La3@[B2@B18]- [18], Td La4[B@B4@B24]0/+/- [19] and Oh La6[La@B24]0/+ [20] have also been predicted in theory. Ln4f atomic orbitals (AOs) are traditionally thought to be contracted and inert. Recent evidence indicates they may play a significant role in the formation of covalent bonding interactions in lanthanide borides, as demonstrated in the experimentally observed planar C2v SmB6- which exhibits effective Sm-B covalent bonding interactions involving obvious participation of the Sm 4f AOs [21]. However, there have been no systematic investigations to date on the ground-state structures and coordination bonding patterns of Ln@B400/+/- . More critically, no quantitative investigations have been made so far to unveil the differences between the Ln 4f AOs and An 5f AOs in coordination bonding capacities in their endohedral metallo-borospherenes.

Keeping the inspirations above in mind, we focus in this work on the ground-state structures and coordination bonding patterns of a series of endohedral Ln@B40^{0/+} complexes. Comprehensive DFT investigations indicate that the ground-state doublet C_{2v} Ce@B40⁺ (²B₁), triplet C_{2v} Ce@B40 (³A₂), quartet C_{2v} Pr@B40 (⁴B₁), quintet C_{2v} Nd@B40 (⁵A₁), sextet C₂ Pm@B40 (⁶A), septet C_{2v} Sm@B40 (⁷A₂), octet D_{2d} Eu@B40 (⁸B₁), and octet C_{2v} Gd@B40⁺ (⁸A₂) possess monotonically decreasing coordination bonding energies in series. The percentage contribution

of the Ln 5d-involved molecular orbitals in these species increase almost monotonically from 66 to 83%, while the Ln 4f-involved molecular orbitals making only minor contributions between 0.3%–12.0%. Detailed bonding analyses unveil the specific coordination bonding patterns of the Ln@B40^{0/+} series and quantitatively reveal the differences in coordination bonding capacities between Ln@B40^{0/+} and An@B40^{0/+/-}.

Computational Methodology

The structures of endohedral Lanthanide-metal-doped Ln@B40^{0/+} (An = Ce, Pr, Nd, Pm, Sm, Eu, Gd) were fully optimized at both the hybrid DFT-PBE0 [22] and DFT-TPSSH [23] levels, with the 6-311+G(d) [24] basis set used for B and the scalar-relativistic Stuttgart energy-consistent pseudopotential with the ECP28MWB_SEG [25] valence basis set employed for Ce-Eu and ECP28MWB_ANO [25–27] used for Gd. The DFT-PBE0 method has proven to be reliable in determining the ground-state structures of lanthanide-metal-doped metallo-borospherenes Ln₃B₁₈⁻ (Ln = La, Tb) in gas-phase photoelectron spectroscopy experiments. [17] Vibrational frequency and wavefunction stability analyses were performed at PBE0 level to make sure that all the low-lying isomers obtained are true minima of the systems without imaginary frequencies. All the PBE0 and TPSSH computations were performed using the Gaussian09 program [28]. Detailed Born-Oppenheimer molecular dynamic (BOMD) simulations were performed on Ce@B40⁺ (1), Ce@B40 (2), Pr@B40 (3), and Eu@B40 (7) at 500 K, Nd@B40 (4), Pm@B40 (5), and C_{2v} Sm@B40 (6) at 300 K, and Gd@B40⁺ (8) at 200 K. BOMD simulation was implemented employing the CP2K [29, 30] code with the GTH-PBE pseudopotentials for B and the TZVP-MOLOPTSR-GTH basis sets for lanthanide atoms, in the NVT ensemble using a Nosé-Hoover thermostat [31]. The infrared and Raman spectra of C_{2v} Ce@B40⁺ (1) and C_{2v} Ce@B40 (2) were simulated at PBE0/6-311+G(d). The UV-vis absorption spectra of Eu@B40 (7) and Gd@B40⁺ (8) were simulated using the time-dependent DFT method (TD-DFT-PBE0) [32, 33]. Chemical bonding patterns were analyzed on the open-shell Ln@B40^{0/+} series employing both the adaptive natural density partitioning (AdNDP) [34, 35] and principal interaction spin orbital (PISO) [36–39] approaches. The PISO analyses were also carried out using the Gaussian 09 program with 6-31G* basis set used for B atom and ECP28MWB_SEG valence basis set employed for Ce-Eu and ECP28MWB_ANO used for Gd element. The VMD [40] program was used for the visualization of structures and molecular orbitals.

Results and Discussions

Structures and Stabilities

The optimized ground states of the doublet C_{2v} Ce@B₄₀⁺ (1, ²B₁), triplet C_{2v} Ce@B₄₀ (2, ³A₂), quartet C_{2v} Pr@B₄₀ (3, ⁴B₁), quintet C_{2v} Nd@B₄₀ (4, ⁵A₁), sextet C_2 Pm@B₄₀ (5, ⁶A), septet C_{2v} Sm@B₄₀ (6, ⁷A₂), octet D_{2d} Eu@B₄₀ (7, ⁸B₁), and octet C_{2v} Gd@B₄₀⁺ (8, ⁸A₂) with slightly distorted or perfect D_{2d} structures are collectively shown in Fig. 1, with alternative low-lying isomers depicted in Fig. S1. Vibrational frequency analyses indicate that the ground states are all true minima on the potential energy surfaces of the systems with the lowest vibrational frequencies of 66.7, 72.1, 75.6, 56.7, 81.2, 23.0, 36.0, and 73.9 cm⁻¹, respectively. The slightly off-centered ground-state doublet C_{2v} Ce@B₄₀⁺ (1, ²B₁) monocation with one unpaired 4f α -electron ($n_\alpha=1$) lies 0.48 (0.52) eV and 2.44 (2.51) eV more stable than the quartet C_{2v} Ce@B₄₀⁺ (⁴A₂) and sextet C_{2v} Ce@B₄₀⁺ (⁶B₂) at PBE0 (TPSSh), respectively (Fig. S1). As expected, neutral C_{2v} Ce@B₄₀ (2, ³A₂) adapts a triplet ground state with two unpaired α -electrons ($n_\alpha=2$) which is 0.67 (0.21) and 0.57 (0.63) eV more stable than the singlet C_{2v} Ce@B₄₀ (¹A₁) and quintet C_s Ce@B₄₀ (⁵A') at PBE0 (TPSSh), respectively (Fig. S1). Detailed BOMD simulations in Fig. S2 indicate that Ce@B₄₀⁺ (1), Ce@B₄₀ (2), Pr@B₄₀ (3), and Eu@B₄₀ (7) are all dynamically stable at 500 K, with the small calculated root-mean-square-deviations of RMSD=0.09, 0.10, 0.11, 0.10 Å and maximum bond length deviations of MAXD=0.31, 0.35, 0.40, 0.35 Å, respectively, while Nd@B₄₀ (4), Pm@B₄₀ (5), and Sm@B₄₀ (6) are stable at 300 K, and Gd@B₄₀⁺ (8) is stable at 200 K, respectively. The optimized neutral Eu@B₄₀ (7, ⁸B₁) appears to be the only species in the endohedral complex series which possesses a perfect D_{2d} geometry, in agreement with the prediction previously

reported in Ref. [21]. It is also worth noticing that octet C_{2v} Gd@B₄₀⁺ (8, ⁸A₂) obtained at PBE0/ECP28MWB_ANO here (Fig. 1 and Fig. S1) possesses a higher symmetry than the previously predicted septet C_s Gd@B₄₀ (⁷A') [21]. Additional calculations indicate that the same ground state of C_{2v} Gd@B₄₀⁺ (8, ⁸A₂) is obtained at PBE0 level with the basis set of ECP28MWB_SEG.

Bonding Pattern Analyses

The high stability of these endohedral metallo-borospherenes originates from their unique coordination bonding patterns. As a typical example with the highest spin multiplicity in the series, Fig. 2 shows the AdNDP and PISO bonding patterns of the octet D_{2d} Eu@B₄₀ (7, ⁸B₁). As indicated in the AdNDP bonding pattern in Fig. 2(a), Eu@B₄₀ (7) possesses 7 singly occupied 1c-1e bonds with the occupation numbers between ON=0.97–1.00 on the Eu coordination center, 40 3c-2e delocalized σ -bonds with ON=1.94–1.96 and 8 6c-2e delocalized σ -bonds with ON=1.84 on the B₄₀ ligand, 4 6c-2e, 4 7c-2e, and 4 8c-2e delocalized π -bonds between the Eu center and B₄₀ ligand with ON=1.68–1.80, and 1 40c-2e σ -bond on the B₄₀ ligand with ON=1.96, indicating that the 12 delocalized π -bonds which all involve the Eu coordination center dominate the coordination bonding interactions between the Eu core and the B₄₀ ligand in the system.

Detailed PISO analyses on Eu@B₄₀ (7) in Fig. 2(b) with the Eu coordination center and B₄₀ ligand as interacting fragments help to provide a more accurate description of the coordination bonding pattern in the complex. It has seven unpaired α -PISO 4f orbitals ($n_\alpha=7$) with the PISO populations of 0.975, 0.975, 0.994, 0.997, 0.997, 0.998, and 0.998, respectively, as the singly occupied molecular

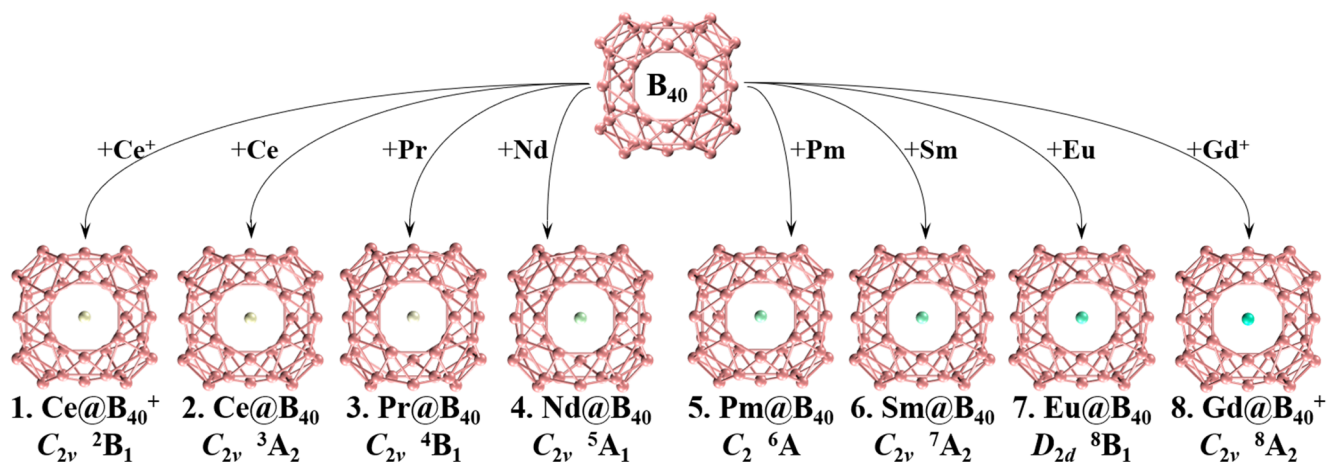


Fig. 1 Optimized ground-state structures of C_{2v} Ce@B₄₀⁺ (1, ²B₁), C_{2v} Ce@B₄₀ (2, ³A₂), C_{2v} Pr@B₄₀⁺ (3, ⁴B₁), C_{2v} Nd@B₄₀ (4, ⁵A₁), C_2 Pm@B₄₀ (5, ⁶A), C_{2v} Sm@B₄₀ (6, ⁷A₂), C_{2v} Eu@B₄₀ (7, ⁸B₁) and

C_{2v} Gd@B₄₀⁺ (8, ⁸A₂) at PBE0 level with the numbers of unpaired α -electrons $n_\alpha=1, 2, 3, 4, 5, 6, 7$, and 7, respectively

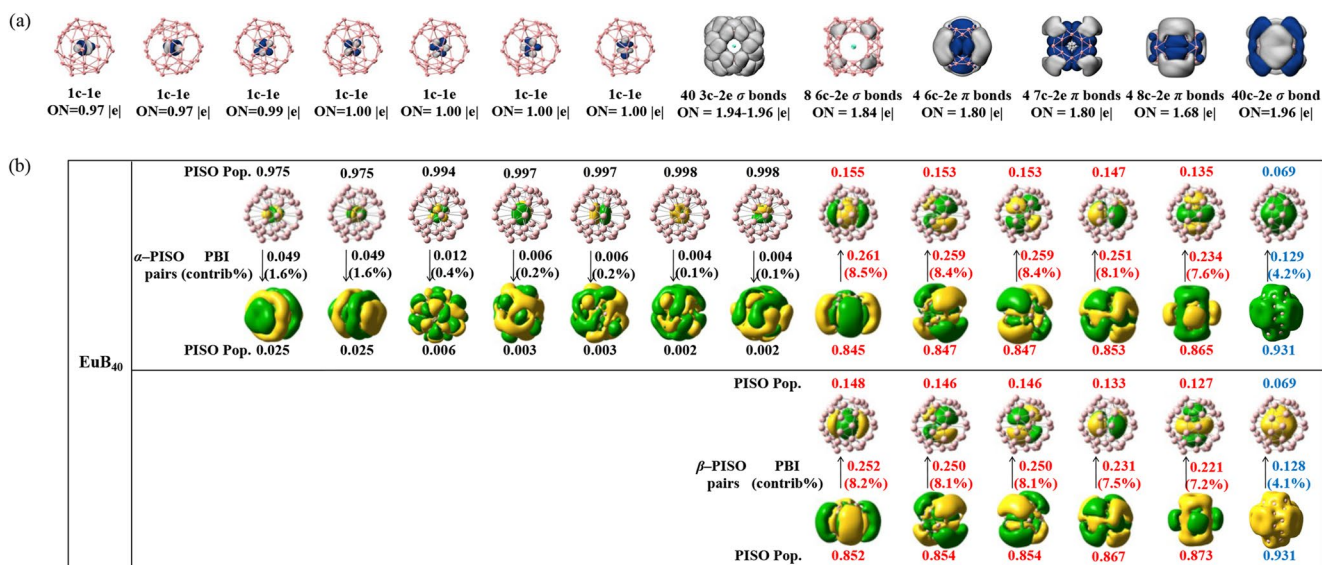


Fig. 2 (a) AdNDP bonding pattern of the octet D_{2d} Eu@B_{40} ($7, {}^8B_1$), with the occupation numbers (ON) indicated. (b) PISO bonding pattern of D_{2d} Eu@B_{40} ($7, {}^8B_1$) using the Eu coordination center and B_{40} ligand as interacting fragments, with the corresponding occupation

numbers (PISO Pop.), PISO-based bond indices (PBI), and percentage contributions (contrib/%) to the overall coordination interactions indicated

orbitals (α -SOMOs) of the complex, which have no the corresponding β -PISO counterparts to correlate with, while all the remaining α -PISO and β -PISO pairs in exact one-to-one corresponding relationships form doubly occupied molecular orbitals, rendering the system an octet ground state (8B_1). The seven unpaired α -PISO 4f orbitals turn out to correspond to the seven 1c-1e bonds obtained by AdNDP analyses in Fig. 2(a), respectively. Their small PISO-based bond indices (PBI) between 0.004 and 0.049 indicate that the interactions between the seven unpaired α -PISO Eu 4f orbitals and the corresponding nearly empty α -molecular orbitals of the B_{40} ligand with the small PISO populations between 0.002 and 0.025 make marginable contributions between 0.1% and 1.6% to the overall coordination energy of the complex. The next five α -PISO pairs and β -PISO pairs highlighted in red which have small but non-negligible PISO populations from Eu 5d AOs (between 0.135 and 0.155) are doubly occupied molecular orbitals which together contribute about 80.1% to the overall coordination energy in the complex, while the correlated α -PISO pair and β -PISO pair which possess the PISO population of 0.069 from Eu 6s atomic orbital in the last column highlighted in blue form another doubly occupied molecular orbital which contributes about 8.3% to the overall coordination energy of the system. Such a PISO bonding pattern clearly indicates that twelve π -coordination interactions between the B_{40} ligand and Eu core mainly contain contributions from the Eu 5d-involved PISOs (80.1% in total), while the Eu 4f-involved PISOs (4.2% in total) and Eu 6s-involved PISO

(8.3% in total) make only minor contributions to the overall coordination interactions in the system.

As shown in Fig. 3 and Fig. S3, Fig. S4, Fig. S5, and Fig. S6, similar AdNDP and PISO bonding patterns exist for doublet C_{2v} Ce@B_{40}^+ ($1, {}^2B_1$), triplet C_{2v} Ce@B_{40} ($2, {}^3A_2$), quartet C_{2v} Pr@B_{40} ($3, {}^4B_1$), quintet C_{2v} Nd@B_{40} ($4, {}^5A_1$), sextet C_2 Pm@B_{40} ($5, {}^6A$), septet Sm@B_{40} ($6, {}^7A_2$), and octet C_{2v} Gd@B_{40}^+ ($8, {}^8A_2$), which have the increasing number of unpaired α -electrons from $n_\alpha = 1, 2, 3, 4, 5, 6$, to 7, respectively. Intriguingly, the first α -SOMO of the sextet Pm@B_{40} ($5, {}^6A$), which contributes 12.1% to the overall coordination interaction, turns out to be a typical α -bond with the comparable Pm 4f α -PISO population of 0.579 and B_{40} α -PISO population of 0.421 and a PISO-based bond indices of $\text{PBI}=0.487$. Such a singly occupied α -PISO pair with non-negligible contributions from both the Pm center and B_{40} ligand corresponds to the 41c-1e covalent bond in the AdNDP bonding pattern shown at bottom in Fig. S5. Similar α -bond exists in quintet Nd@B_{40} ($4, {}^5A_1$) which contributes 5.4% to the overall coordination interaction with the Nd 4f α -PISO population of 0.138 and B_{40} α -PISO population of 0.862 (Fig. 4). Thus, Ln 4f AOs make obvious contributions to the singly occupied α -bonds in both Nd@B_{40} ($4, {}^5A_1$) and Pm@B_{40} ($5, {}^6A$), analogous to the situation observed in SmB_6^- which exhibits effective Sm-B covalent bonding interactions involving obvious participation of the Sm 4f AOs. [1] More interestingly, the first singly occupied α -SOMO of Ce@B_{40} (2) which possesses the marginal Ce 4f α -PISO population of 0.947 corresponds to the 40c-1e

Fig. 3 Singly occupied PISO α -orbitals of Ce@B₄₀⁺ (C_{2v} , ²B₁), Ce@B₄₀ (C_{2v} , ³A₂), Pr@B₄₀ (C_{2v} , ⁴B₁), Nd@B₄₀ (C_2 , ⁵A₁), Pm@B₄₀ (C_2 , ⁶A), Sm@B₄₀ (C_{2v} , ⁷A₂), Eu@B₄₀ (D_{2d} , ⁸B₁), and Gd@B₄₀⁺ (C_{2v} , ⁸A₂) with the α -spin populations (PISO Pop.) associated with the principal interacting spin orbitals, PISO-based bond indices (PBI), and their percentage contributions to the overall coordination interactions between the lanthanide coordination center and B₄₀ ligand indicated. The corresponding singly occupied 1c-1e bonds in these species obtained by AdNDP analyses are shown at the bottom, with the occupation numbers (ON) indicated

Figure 1 consists of two plots. The left plot shows the overall coordination energy (blue line with circles) and the percentage contribution of f, d, and s orbitals (red lines with various markers) for Ce@B₄₀ and Ln@B₄₀ complexes. The right plot shows the percentage contribution of f, d, and s orbitals for Ln@B₄₀ complexes. Both plots show a general trend of decreasing coordination energy and increasing orbital contribution as the lanthanide series progresses from Ce to Gd.

Left Plot: Coordination Energy and Orbital Contribution for Ce@B₄₀ and Ln@B₄₀ Complexes

Complex	Ln	Overall Coordination Energy (eV)	f-orbital Contribution (%)	d-orbital Contribution (%)	s-orbital Contribution (%)
Ce@B ₄₀ ⁺	1	~82	~10	~10	~80
Ce@B ₄₀	2	~78	~10	~10	~80
Pr@B ₄₀	3	~62	~8	~10	~82
Nd@B ₄₀	4	~48	~8	~10	~82
Pm@B ₄₀	5	~46	~12	~10	~78
Sm@B ₄₀	6	~42	~10	~10	~80
Eu@B ₄₀	7	~38	~10	~10	~80
Gd@B ₄₀	7	~10	~10	~10	~80

Right Plot: Orbital Contribution for Ln@B₄₀ Complexes

Complex	Ln	f-orbital Contribution (%)	d-orbital Contribution (%)	s-orbital Contribution (%)
Ce@B ₄₀ ⁺	1	~8	~66	~26
Ce@B ₄₀	2	~8	~66	~26
Pr@B ₄₀	3	~3	~69	~28
Nd@B ₄₀	4	~5	~72	~23
Pm@B ₄₀	5	~12	~70	~18
Sm@B ₄₀	6	~7	~75	~18
Eu@B ₄₀	7	~4	~80	~16
Gd@B ₄₀	7	~1	~83	~16

σ -bond on the B₄₀ ligand with ON=0.91, while the first singly occupied α -SOMO of quartet Pr@B₄₀ (**3**) which has the minor Pr 4f α -PISO population of 0.077 and predominating B₄₀ α -PISO population of 0.923 represents the

lanthanide center to B_{40} ligand, with the corresponding electrostatic interactions contributing 2.1% and 3.1% to the overall coordination energies of the corresponding systems, respectively. The results obtained above suggest that, relative to the weak ionic interactions of $Ce^+@B_{40}^-$ (2) and $Pr^+@B_{40}^-$ (3) in PISO approach, covalent interactions predominate the coordination interactions in these lanthanide-doped metallo-borospherene complexes.

Percentage Contributions of Ln 4f-, 5d-, and 6s-involved PISO Pairs To the Total Coordination Interactions

In order to compare the percentage contributions of Ln 4f-, 5d-, and 6s-involved PISO pairs to the overall Ln- B_{40} coordination interaction energies, we categorized the orbital types of Ln atoms involved in the PISO bonding patterns by their orbital shapes and considered the contributions of the corresponding PISO pairs separately. As illustrated in Fig. 4(a), as the number of unpaired α -electrons with parallel spins increases from $n_\alpha = 1, 2, 3, 4, 5, 6, 7$ and 7 in the $Ce@B_{40}^+$ (1), $Ce@B_{40}$ (2), $Pr@B_{40}$ (3), $Nd@B_{40}$ (4), $Pm@B_{40}$ (5), $Sm@B_{40}$ (6), $Eu@B_{40}$ (7), and $Gd@B_{40}^+$ (8) series, the calculated overall Ln- B_{40} coordination interaction energies decrease monotonically from $E_c = 7.22, 6.93, 5.67, 4.85, 4.67, 4.29, 4.02$, to 2.07 eV, respectively. In percentage contributions, the dominating contributions of the Ln 5d-involved PISO pairs to the overall E_c increase almost monotonically from 66.1%, 66.2%, 69.9%, 71.8%, 69.7%, 75.4%, 80.0–83.4%, while the corresponding minor contributions of the Ln 4f-involved PISO pairs vary from 8.8%, 7.7%, 3.1%, 5.4%, 12.1%, 6.8%, 4.2%, to 0.3% and that of Ln 6s-involved PISO pairs remain basically unchanged in the narrow range from 6.5%, 6.5%, 6.5%, 6.6%, 6.9%, 7.9%, 8.3%, to 7.1%.

As shown in Fig. 4(b), PISO analyses also provide a quantitative way to identify the difference in coordination bonding patterns between lanthanide-doped borospherenes highlighted in red and actinide-doped endohedral borospherenes highlighted in purple. [22] With the numbers of the unpaired α -electrons increasing from $n_\alpha = 1$ to 7, the contributions of Ln 5d-involved PISO pairs in $Ln@B_{40}^{+/0}$ increase monotonically from 66 to 83% and the contributions of An 6d-involved PISO pairs $An@B_{40}^{+/0/-}$ increase monotonically from 47–72% [22], while the contributions of Ln 4f-involved PISO pairs vary between 0.3 and 12.0% and the contributions of the An 5f-involved PISO pairs decrease monotonically from 41% to 1%. In average, the dominating Ln 5d-involved PISO pairs contribute about 72.8% to the overall coordination bonding energies in $Ln@B_{40}^{+/0}$, 19.3% higher than that (53.5%) of the An 6d-involved PISO pairs

in $An@B_{40}^{+/0/-}$ [22], while the minor 4f-involved PISO pairs in $Ln@B_{40}$ contribute about 6.1% to the overall E_c , 15.7% lower than that (21.8%) of the 5f-involved PISOs in $An@B_{40}^{+/0/-}$. These observations agree with the fundamental electronic structure differences between lanthanides and actinides, where the Ln 4f orbitals in lanthanides exhibit less spatial extension and weaker bonding capability than the An 5f orbitals in actinides, while the Ln 5d orbitals in the former exhibit greater spatial extension and stronger bonding capability than the An 6d orbitals in actinides. Both Ln 6s-involved and An 7s-involved PISO pairs appear to make minor contributions to the overall coordination interaction energies in their boride complexes.

Simulated IR, Raman and UV-Vis Spectra

The IR and Raman spectra of C_{2v} $Ce@B_{40}^+$ (1, 2B_1) and C_{2v} $Gd@B_{40}^+$ (8, 8A_2) are depicted in Fig. 5(a) at PBE0 level to facilitate their future experimental characterizations. $Ce@B_{40}^+$ (1) exhibits four distinct IR absorption peaks at 725 (b_2), 808 (a_1), 1119 (b_2), and 1284 (b_2) cm^{-1} , corresponding to the tangential vibrational modes b_2, a_1, b_2 , and b_2 of the B_{40} boron skeleton, respectively, respectively [23] In the Raman spectrum, characteristic vibrational modes are identified at 247 (b_1), 277 (a_2), 454 (a_1), and 635 (a_1) cm^{-1} . Similar IR and Raman spectra features are obtained for $Gd@B_{40}^+$ (8). Notably, $Ce@B_{40}^+$ (1) and $Gd@B_{40}^+$ (8) exhibit characteristic radial breathing modes (RBMs) at 454 cm^{-1} (a_1) and 463 cm^{-1} (a_1), respectively, which are slightly blue-shifted compared to the RBM of the empty D_{2d} B_{40} borospherene (428 cm^{-1} (a_1)) at the same theoretical level. The UV-Vis spectra of neutral C_{2v} $Ce@B_{40}$ (2) and D_{2d} $Eu@B_{40}$ (7) are also computationally simulated in Fig. 5(b) using TD-DFT-PBE0 approach with 200 excited states included. For $Ce@B_{40}$ (2), the main adsorption peaks occur at 344 (3A_1), 388 (3A_1), 404 (3A_1), and 559 (3B_2) nm, while for $Eu@B_{40}$ (7), the primary absorption peaks are predicted at 366 (8B_2), 399 (8E), 448 (8B_2), and 659 (8E) nm.

Conclusion

We have predicted in this work the ground-state structures of doublet $Ce@B_{40}^+$ (1), triplet $Ce@B_{40}$ (2), quartet $Pr@B_{40}$ (3), quintet $Nd@B_{40}$ (4), sextet $Pm@B_{40}$ (5), septet $Sm@B_{40}$ (6), octet $Eu@B_{40}$ (7), and octet $Gd@B_{40}^+$ (8) at both PBE0 and TPSSh levels, revealed their coordination bonding patterns using both the PISO and AdNDP approaches, and calculated the percentage contributions of Ln 4f-, 5d-, and 6s-involved PISO pairs to the overall coordination interaction energies at PBE0, deciphering

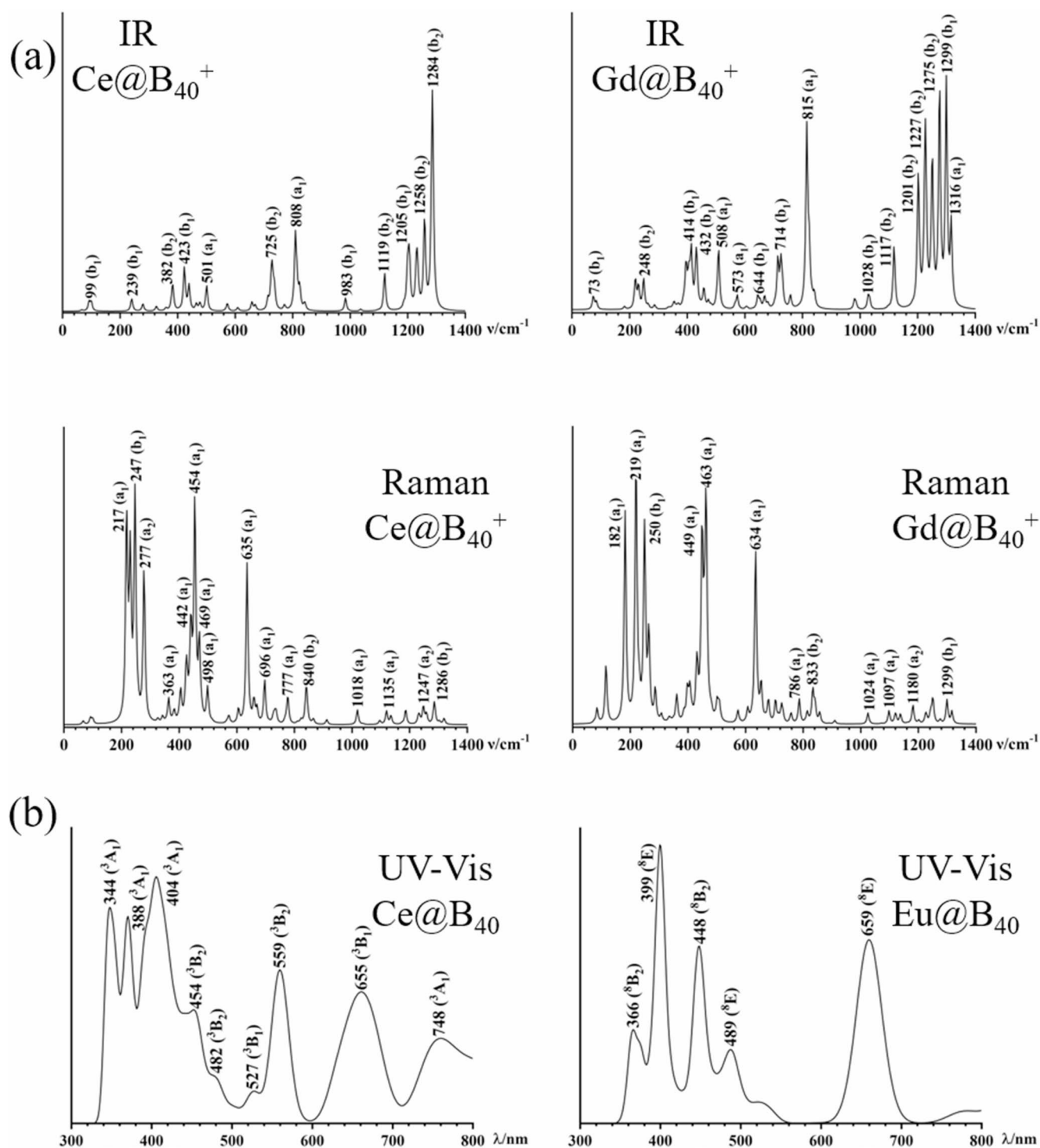


Fig. 5 Simulated (a) IR and Raman spectra of C_{2v} $\text{Ce}@\text{B}_{40}^{+}$ (**1**, 2B_1) and C_{2v} $\text{Gd}@\text{B}_{40}^{+}$ (**8**, 8A_2) and (b) UV-Vis spectra of C_{2v} $\text{Ce}@\text{B}_{40}$ (**2**, 3A_2) and D_{2d} $\text{Eu}@\text{B}_{40}$ (**7**, 8B_1) at PBE0 level

the coordination bonding nature of these lanthanide-doped endohedral metallo-borospherenes both qualitatively and quantitatively. In average, the minor Ln 4f-involved PISOs in $\text{Ln}@\text{B}_{40}$ contribute about 15.7% less than that of the An 5f-involved PISOs in $\text{An}@\text{B}_{40}^{+/0/-}$, while the dominating

Ln 5d-involved PISOs in $\text{Ln}@\text{B}_{40}$ contribute 19.3% more than that of the An 6d-involved PISOs in $\text{An}@\text{B}_{40}^{+/0/-}$. Ln 4f AOs are found to make obvious contributions to the singly occupied α -bonds in both $\text{Nd}@\text{B}_{40}$ (**4**) and $\text{Pm}@\text{B}_{40}$ (**5**) which belong to covalent bonding interactions between the

Ln center and B₄₀ ligand. Such high-spin lanthanide-doped endohedral metallo-borospherenes could be realized in future experiments to form various nanoclusters and nanocrystals which may have wide potential applications in both chemistry and materials science.

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Author Contributions S-D Li designed the project and finalized the manuscript and X-N Zhao, T Zhang and X-Q Lu performed the calculations. All the authors participated in the discussion and preparation of the manuscript.

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Data Availability No datasets were generated or analysed during the current study.

Declarations

Competing Interests The authors declare no competing interests.

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Authors and Affiliations

Xiao-Ni Zhao¹ · Ting Zhang¹ · Xiao-Qin Lu² · Si-Dian Li¹

✉ Si-Dian Li
lisidian@sxu.edu.cn

¹ Nanocluster Laboratory, Institute of Molecular Science, Shanxi University, Taiyuan, Shanxi 030006, China

² Shanxi Center for Testing of Functional Agro-Products, Shanxi Agricultural University, Taiyuan 030031 Shanxi, China