

# Predicting the Structural Transition in Medium-Sized Boron Nanoclusters: From Bilayer $B_{64}$ , $B_{66}$ , $B_{68}$ , $B_{70}$ , and $B_{72}$ to Core-Shell $B_{74}$

Ling Pei<sup>+</sup>,<sup>[a, b]</sup> Qiao-Qiao Yan<sup>+</sup>,<sup>[a]</sup> and Si-Dian Li<sup>\*[a]</sup>

Recent experimental observation of the first bilayer clusters  $B_{48}^{-/0}$  reveals a new structural domain in boron nanostructures. Inspired by the previously reported bilayer  $B_{48}$ ,  $B_{54}$ ,  $B_{60}$ , and  $B_{62}$  and based on extensive global-minimum searches and density-functional theory calculations, we predict herein a new series of medium-sized bilayer boron nanoclusters  $C_2 B_{64}$  (I),  $D_2 B_{66}$  (II),  $D_2 B_{68}$  (III),  $C_1 B_{70}$  (IV), and  $C_1 B_{72}$  (V) which all contain an elongated  $B_{46}$  bilayer hexagonal prism at the center with four effective interlayer 2c–2e B–B  $\sigma$  bonds formed between the top and bottom layers and the bilayer to core-shell structural transition at  $B_{74}$  where core-shell species start to dominate in thermody-

namics, defining the up-limit of the bilayer boron nanoclusters at  $B_{72}$ . The newly obtained bilayer  $C_2 B_{64}$  (I),  $D_2 B_{68}$  (III), and  $C_1 B_{70}$  (IV) appear to be systematically more stable than the previously reported cage-like  $D_{4d} B_{64r}$  core-shell  $C_1 B_{68r}$  and quasi-planar  $C_{3v} B_{70r}$ , respectively. Detailed bonding analyses indicate that these bilayer species follow the universal bonding pattern of  $\sigma + \pi$  double delocalization, rendering three-dimensional aromaticity to the systems. The IR, Raman, and UV-vis spectra of the concerned bilayer species are computationally simulated to facilitate their future characterizations.

## Introduction

As a prototypical electron-deficient element, boron has a rich chemistry<sup>[1]</sup> next only to carbon in the periodical table. Persistent joint photoelectron spectroscopy (PES) and first-principles theory investigations in the past two decades have unveiled a rich landscape for size-selected  $B_n^{-/0}$  boron clusters from planar or quasi-planar (2D) boron sheets ( $n=3-38$ , 41–42),<sup>[2]</sup> cage-like borospherenes  $D_{2d} B_{40}^{-/0}$  and  $C_2/C_2 B_{39}^{-}$ ,<sup>[3]</sup> to bilayer  $D_{2h} B_{48}^{-/0}$ .<sup>[4]</sup> The borospherene family has been extended at first-principles theory level to include the  $B_n^q$  series ( $n=36-42$ ,  $q=n-40$ ) which are all composed of twelve interwoven boron double chains with six hexagons or heptagons on the cage surface in a universal bonding pattern of  $\sigma + \pi$  double delocalization.<sup>[3,5]</sup> Seashell-like borospherenes  $C_2 B_{28}^-$  and  $C_5 B_{29}^-$  were late observed in PES measurements as minor isomers competing with their 2D global minimum (GM) counterparts.<sup>[6]</sup> Following the same structural motif, seashell-like  $C_5 B_{29}^+$ ,  $C_2 B_{31}^+$ ,  $C_2 B_{32r}$ ,  $C_2 B_{34r}$ ,  $C_2 B_{35}^+$ , and  $C_2 B_{38}^+$  were predicted in theory recently.<sup>[7]</sup> Boron cluster monocations  $B_n^+$  ( $n=16-25$ ) were shown to possess double-ring tubular geometries in a joint ion-mobility measurements and density functional theory (DFT)

investigation.<sup>[8]</sup> Neutral core-shell  $B_n$  clusters  $C_1 B_{68r}$ ,  $C_1 B_{74r}$ ,  $C_1 B_{80r}$ ,  $D_{3d} B_{98r}$ ,  $C_{2h} B_{100r}$ ,  $C_5 B_{101r}$ ,  $D_{3d} B_{102r}$ ,  $C_5 B_{111r}$ ,  $C_5 B_{112r}$ ,  $C_5 B_{113r}$ , and  $C_5 B_{114r}$  with a partially or completely encapsulated icosahedral  $B_{12}$  core were predicted to dominate in thermodynamics in a large cluster size range starting from  $n=68$ , with  $C_5 B_{112}$  being the most stable core-shell species in thermodynamics reported to date between  $B_{68}-B_{130r}$ <sup>[9]</sup> indicating that an icosahedral  $B_{12}$  cage serves as the energetically most favorable “seeds” to form three-dimensional (3D) boron allotropes in bottom-up approaches.  $B_{84}$  as an exception was previously predicted to favor a quasi-planar  $C_{2v}$  geometry at DFT level.<sup>[10]</sup> Recent extensive GM searches showed that complicated structural competitions exist in medium-sized  $B_n$  clusters, with  $C_2 B_{46}$  being the smallest core-shell boron cluster ( $B_4@B_{42}$ ) reported to date and  $D_{2h} B_{48r}$ ,  $C_2 B_{54r}$ ,  $C_{2h} B_{60r}$ , and  $C_1 B_{62}$  being the first bilayer boron clusters with a  $B_{38}$  bilayer hexagonal prism at the center surrounded by certain numbers of boron atoms on the waist.<sup>[9f,11]</sup> Encouragingly, bilayer  $D_{2h} B_{48}^{-/0}$  clusters were very recently observed in a joint PES and first-principles theory investigation, providing the first experimental evidence for the viability of free-standing 2D bilayer borophenes similar to monolayer graphenes.<sup>[4]</sup> However, there has been no theoretical or experimental evidence reported to date on bilayer boron clusters beyond  $B_{62}$  where bilayer structures prevail over other geometries and the up-limit of bilayer boron nanoclusters at which the bilayer to core-shell structural transition occurs still remains unknown.

Based on extensive GM searches and first-principles theory calculations, we predict in this work the bilayer  $C_2 B_{64}$  (I),  $D_2 B_{66}$  (II),  $D_2 B_{68}$  (III),  $C_1 B_{70}$  (IV), and  $C_1 B_{72}$  (V) as the GMs of the systems beyond  $B_{62}$  and the bilayer to core-shell structural transition at  $B_{74}$  where the most stable bilayer isomer obtained is found to be slightly less stable than the previously reported core-shell  $C_1 B_{74}$ <sup>[9d]</sup> in thermodynamics. The bilayer  $B_{64r}$ ,  $B_{66r}$ ,  $B_{68r}$ ,

[a] Dr. L. Pei,<sup>+</sup> Dr. Q.-Q. Yan,<sup>+</sup> Prof. Dr. S.-D. Li  
Institute of Molecular Science,  
Shanxi University,  
Taiyuan 030006, China  
E-mail: lisidian@sxu.edu.cn

[b] Dr. L. Pei<sup>+</sup>  
Department of Chemical Engineering and Safety,  
Binzhou University,  
Binzhou 256603, China

[\*] L. Pei and Q.-Q. Yan contributed equally to this work.

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$B_{70r}$  and  $B_{72}$  all contain an elongated  $B_{46}$  bilayer hexagonal prism at the center surrounded by certain numbers of boron atoms on the waist, featuring four effective interlayer B–B  $\sigma$  bonds between eight inward-buckled boron atoms on the top and bottom layers. These highly stable bilayer nanoclusters follow the universal bonding pattern of  $\sigma + \pi$  double delocalization and are three dimensionally aromatic in nature.

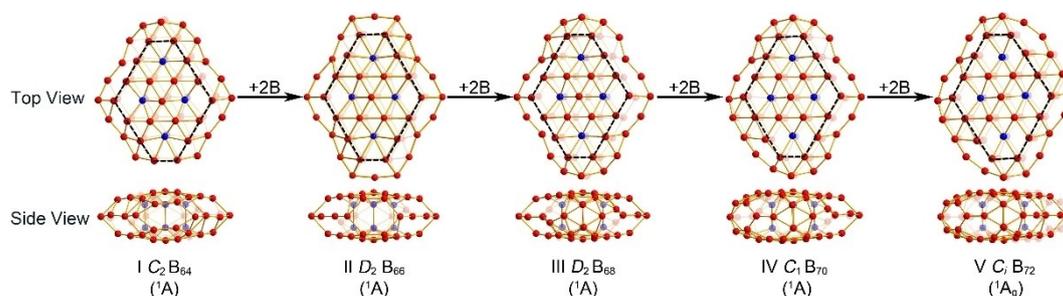
## Results and Discussions

### Structures and Stabilities

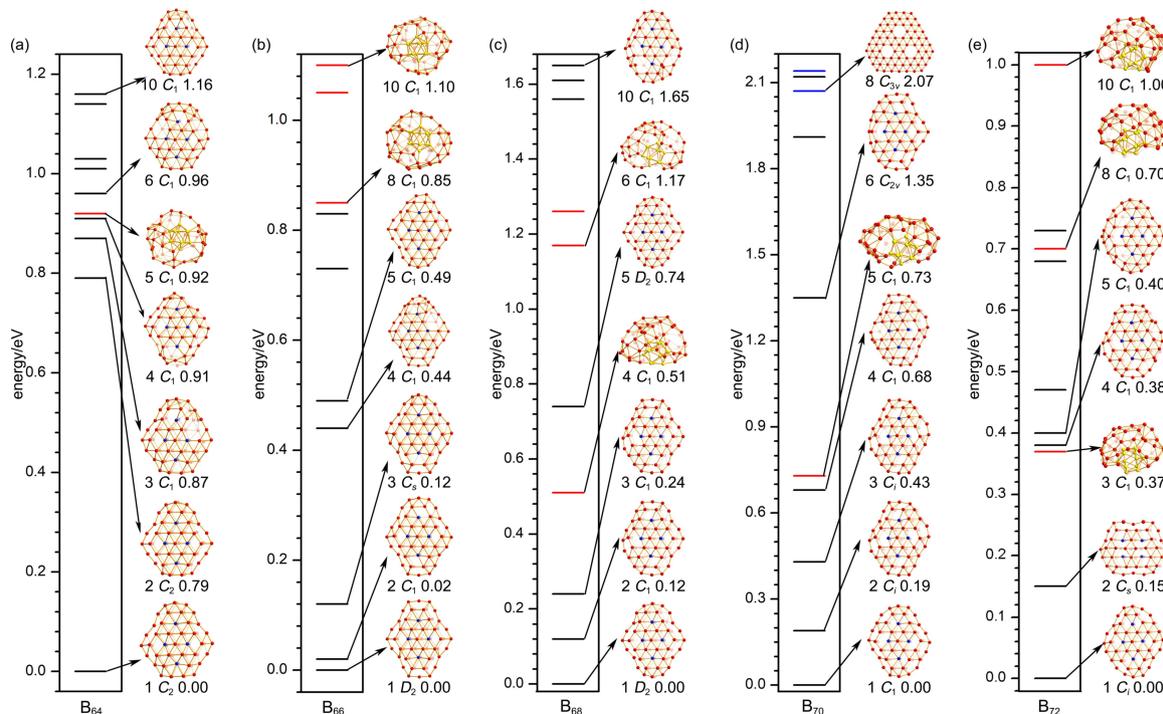
The optimized bilayer  $C_2 B_{64}$  (I),  $D_2 B_{66}$  (II),  $D_2 B_{68}$  (III),  $C_1 B_{70}$  (IV), and  $C_1 B_{72}$  (V) GMs are shown in Figure 1, with more alternative

low-lying isomers depicted in Figure S1. The configurational energy spectra of  $B_{64r}$ ,  $B_{66r}$ ,  $B_{68r}$ ,  $B_{70r}$  and  $B_{72}$  are depicted in Figure 2 at PBE0/6-311 + G(d) level. The calculated lowest-lying vibrational frequencies  $\nu_{\min}$  at PBE0/6-31G(d) and HOMO-LUMO energy gaps  $\Delta E_{\text{gap}}$ , cohesive energy per atom  $E_{\text{coh}}$ , and nucleus-independent chemical shifts (NICS)<sup>[12]</sup> at PBE0/6-311 + G(d)<sup>[13]</sup> are comparatively tabulated in Table 1 for bilayer  $B_n$  clusters in the size range between  $n=48-72$ . The  $E_{\text{coh}} \sim n$  curves of  $B_n$  clusters ( $n=48-74$ ) in different structural motifs are compared in Figure 3.

As shown in Figure S1a in the ESI†, eighteen out of the twenty lowest-lying isomers of  $B_{64}$  within 1.74 eV at PBE0/6-31G(d) all possess bilayer geometries. The axially chiral bilayer  $C_2 B_{64}$  (I) which can be obtained by adding a  $B_4$  rhombus over the  $B_6$  hexagonal window on the upper end of the previously reported



**Figure 1.** Top and side views of the optimized structures of  $C_2 B_{64}$  (I),  $D_2 B_{66}$  (II),  $D_2 B_{68}$  (III),  $C_1 B_{70}$  (IV), and  $C_1 B_{72}$  (V) at PBE0/6-311 + G(d) level. The elongated  $B_{46}$  bilayer hexagonal prisms at the center are highlighted in black circles and inward-buckled B atoms interconnected by four interlayer B–B bonds are colored in blue.



**Figure 2.** Configurational energy spectra of (a)  $B_{64r}$ , (b)  $B_{66r}$ , (c)  $B_{68r}$ , (d)  $B_{70r}$  and (e)  $B_{72}$  at PBE0/6-311 + G(d) level. Black, red and blue horizontal lines represent bilayer, core-shell and quasi-planar structures, respectively. Relative energies are given in eV.

**Table 1.** Comparison of the  $\nu_{\min}$  ( $\text{cm}^{-1}$ ),  $\Delta E_{\text{gap}}$  (eV),  $E_{\text{coh}}$  (eV/atom) and NICS (ppm) values calculated for the bilayer  $B_n$  ( $n=48-72$ ) clusters reported to date.

$B_n$	$\nu_{\min}$ [ $\text{cm}^{-1}$ ] <sup>[a]</sup>	$\Delta E_{\text{gap}}$ [eV] <sup>[b]</sup>	$E_{\text{coh}}$ (eV/atom) <sup>[b]</sup>	NICS [ppm] <sup>[b]</sup>
$D_{2h} B_{48}$	75.11	1.49	-5.3433	-31.2
$C_2 B_{54}$	128.4	1.98	-5.3653	-33.2
$C_{2h} B_{60}$	105.5	1.14	-5.3646	-41.5
$C_1 B_{62}$	135.5	1.45	-5.3824	-31.4
$C_2 B_{64}$	110.7	1.54	-5.3858	-20.5
$D_2 B_{66}$	18.8	1.13	-5.3920	-33.3
$D_2 B_{68}$	137.5	1.47	-5.4025	-16.8
$C_1 B_{70}$	22.0	1.43	-5.4103	-12.3
$C_i B_{72}$	107.6	1.55	-5.4132	-5.7

[a] At PBE0/6-31G(d) level [b] At PBE0/6-311 + G(d) level

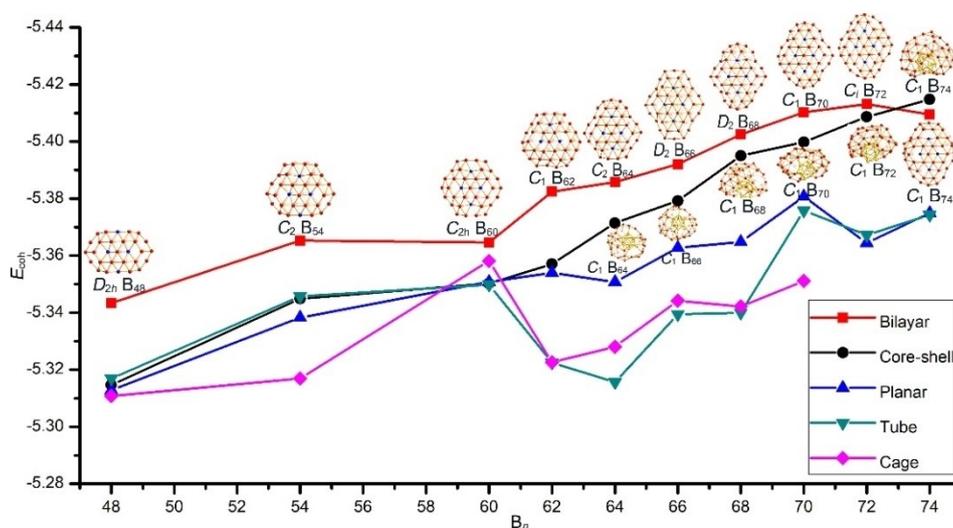
$C_{2h} B_{60}$ <sup>[11]</sup> contains an elongated  $B_{46}$  bilayer hexagonal prism at the center (highlighted in a black circle in Figure 1) surrounded by 18 B atoms on the waist. It features four effective interlayer B–B  $\sigma$  bonds between eight inward-buckled boron atoms (highlighted in blue) on the top and bottom layers.  $C_2 B_{64}$  (I) appears to be the well defined GM of  $B_{64}$  strongly favored in thermodynamics over other isomers (Figure 2a): it is 0.79 eV and 0.92 eV more stable than the second lowest-lying bilayer  $C_2 B_{64}$  and first lowest-lying core-shell  $C_1 B_{64}$  at PBE0/6-311 + G(d), respectively (Figure S1a). The previously proposed cage-like  $D_{4d} B_{64}$ <sup>[14]</sup> proves to be much less stable than the  $C_2$  GM (by 3.53 eV) at PBE0/6-31G(d) (Figure S1a).

The potential energy surface of  $B_{66}$  appears to be more congested (Figure 2b). Its seven lowest-lying isomers within 0.85 eV at PBE0/6-311 + G(d) all possess bilayer configurations, with the axially chiral GM  $D_2 B_{66}$  (II) which can be obtained from the previously reported  $C_{2h} B_{60}$ <sup>[11]</sup> by adding two –B–B– chains over two  $B_6$  hexagons on the upper and down ends being only 0.02 and 0.15 eV more stable than the second lowest-lying isomer bilayer  $C_1 B_{66}$  at PBE0/6-311 + G(d) and

TPSSH/6-311 + G(d)<sup>[13a,15]</sup> (Figure S1b), respectively. These two close-lying bilayer isomers are practically iso-energetic in thermodynamics and may coexist in gas-phase experiments. The prototypical core-shell  $C_1 B_{66r}$  quasi-planar  $C_3 B_{66r}$  cage-like  $D_2 B_{66r}$  and triple-ring tubular  $C_{2h} B_{66}$ <sup>[16]</sup> are found to be 0.84, 2.24, 3.43, and 4.15 eV less stable than  $B_{66}$  (II) at PBE0/6-31G(d), respectively (Figure S1b).

$B_{68}$  is an object more concerned in this work. It has a complicated configurational competition between bilayer and core-shell isomers, with the core-shell isomers starting to appear as the fourth, sixth, and seventh lowest-lying isomers (Figure 2c and Figure S1c). An amorphous incomplete core-shell  $C_1 B_{68}$  with a partially encapsulated icosahedral  $B_{12}$  core was proposed by Zhao and co-workers in 2010 via unbiased GM searches.<sup>[9d]</sup> However, in conjunction with manual structural constructions based on known bilayer species, extensive GM searches performed in this work indicate that the high-symmetry axially chiral bilayer  $D_2 B_{68}$  (III) is 0.51 eV and 0.58 eV more stable than the previously reported core-shell  $C_1 B_{68}$ <sup>[9d]</sup> at PBE0/6-311 + G(d) and TPSSH/6-311 + G(d), respectively (Figure S1c), presenting a bilayer  $D_2 B_{68}$  as the GM of the system for future experiments to confirm. The second and third lowest-lying bilayer isomers also appear to be obviously more stable than core-shell  $C_1 B_{68}$  (Figure 3c).  $D_2 B_{68}$  (III) can be obtained either from the previously reported  $C_{2h} B_{60}$ <sup>[11]</sup> by adding two equivalent  $B_4$  rhombuses over two  $B_6$  hexagons on the upper and down ends or from  $D_2 B_{66}$  (II) by adding two penta-coordinate  $\eta^5\text{-B}$  atoms at the two ends. It possesses an elongated  $B_{46}$  bilayer hexagonal prism at the center sealed by 22 B atoms on the waist, featuring four effective interlayer B–B  $\sigma$  bonds formed between the top and bottom layers. The prototypical quasi-planar  $C_1 B_{68r}$  cage-like  $C_1 B_{68r}$  and hexa-ring tubular  $C_1 B_{68}$  are found to lie 2.98, 4.57, 4.57 eV higher than the bilayer  $D_2$  GM at PBE0/6-31G(d), respectively (Figure S1c).

Adding two more boron atoms to  $D_2 B_{68}$  (III) to form a  $B_6$  hexagonal window on one side of the waist generates the most



**Figure 3.** Binding energy per atom of the  $B_n$  clusters ( $n=48-74$ ) in different structural motifs: bilayer (squares), core-shell (circles), planar (triangles), tubular (inverted triangles), and cage-like (rhombuses).

stable bilayer  $C_1 B_{70}$  (IV) which lies 2.07 and 1.39 eV more stable than the previously predicted quasi-planar triplet  $C_{3v} B_{70}$  ( $^3A_1$ )<sup>[10]</sup> at PBE0/6-311+G(d) and TPSSh/6-311+G(d), respectively (Figure S1d). Five more bilayer isomers (the second, third, fourth, sixth, and seventh isomers) and one core-shell isomer (the fifth isomer) are found to be energetically more favorable than quasi-planar  $C_{3v} B_{70}$ . The tubular  $C_5 B_{70}$  is found to lie 2.81 eV higher than  $B_{70}$  (IV) at PBE0/6-31G(d) (Figure S1d).

Addition of two more boron atoms to  $B_{70}$  generates more complicated structural competitions to the system. As shown in Figure 2e, bilayer  $C_i B_{72}$  (V) is the most stable isomer obtained for  $B_{72}$  in this work, with both the first ( $C_1$ ) and second ( $C_2$ ) lowest-lying core-shell isomers being 0.37 and 0.70 eV less stable than the bilayer GM at PBE0/6-311+G(d), respectively.  $B_{72}$  (V) can be obtained either from  $C_1 B_{70}$  (IV) by adding two more boron atoms to the system to form two symmetrically distributed  $B_6$  hexagonal windows on two diagonal edges or from  $D_2 B_{68}$  (III) by adding four boron atoms to the system to form two equivalent  $B_6$  hexagonal windows on two diagonal edges. Typical tubular and quasi-planar isomers appear to be at least 3.64 eV less stable than  $B_{72}$  (V) at PBE0/6-31G(d) (Figure S1e). It is noticed that  $B_{72}$  (V) possesses the highest cohesive energy per atom of  $E_{coh} = -5.4132$  eV in the bilayer  $B_n$  series between  $n = 48-72$  (Table 1 and Figure 3) where, in overall, the bilayer species prevail in relative energies over planar, core-shell, and tubular structural motifs (Figure 3).

With two more boron atoms added in, a dramatic bilayer to core-shell structural transition occurs at  $B_{74}$  where the previously reported incomplete core-shell  $C_1 B_{74}$ <sup>[9d]</sup> appears to be slightly more stable than the most stable bilayer  $C_1 B_{74}$  obtained in this work by 0.40 eV and 0.27 eV at PBE0/6-311+G(d) and TPSSh/6-311+G(d) levels, respectively (Figure S1f). This observation defines  $C_i B_{72}$  (V) as the up-limit of the bilayer structural motif in medium-sized  $B_n$  clusters beyond which core-shell structures start to dominate in thermodynamics (Figure 3). The smallest core-shell  $B_n$  cluster is thus  $B_{74}$  at both PBE0 and TPSSh levels, instead of the previously predicted  $B_{68}$ <sup>[9d]</sup> with both  $B_{70}$  and  $B_{72}$  possessing bilayer GMs which are obviously more stable than their core-shell counterparts (Figure 2 and 3).

Bilayer  $B_{64}$  (I),  $B_{66}$  (II),  $B_{68}$  (III),  $B_{70}$  (IV), and  $B_{72}$  (V) with the large calculated HOMO-LUMO gaps of 1.54, 1.13, 1.47, 1.43, and 1.55 eV (Table 1) also appear to be dynamically stable, as demonstrated in Figure S2 for  $D_2 B_{66}$  (II) and  $D_2 B_{68}$  (III) which have the small average root-mean-square-deviations of RMSD = 0.13, 0.20 Å and maximum bond length deviations of MAXD = 0.66, 1.33 Å at 700 K, respectively (Figure S2). Similar situations exist in other bilayer species. These bilayer species which can be viewed as squashed cage-like structures all appear to be 3D aromatic in nature, as evidenced by their negative calculated NICS values NICS =  $-5.7 \sim -33.3$  ppm in the size range between  $n = 64-72$  (Table 1).

## Bonding Analyses

To better interpret the high stabilities of these bilayer species, detailed AdNDP<sup>[17]</sup> bonding analyses were performed on the

high-symmetry  $D_2 B_{66}$  (II) and  $D_2 B_{68}$  (III) in Figure 4. As shown in Figure 4a,  $B_{66}$  (II) possesses 4 interlayer 2c-2e B-B  $\sigma$ -bonds between eight inward-buckled B atoms on the top and bottom layers with the occupation numbers of ON = 1.80-1.87 |e|, 28 2c-2e B-B  $\sigma$ -bonds on the waist with ON = 1.73-1.83 |e|, 40 3c-2e  $\sigma$  bonds on the top and bottom layers with ON = 1.78-1.93 |e|, 2 5c-2e  $\sigma$  bonds at two corners on the left and right with ON = 1.73 |e|, 6 9c-2e  $\sigma$  bonds on the upper and down ends with ON = 1.85-1.95 |e|, and 1 14c-2e  $\sigma$  bond at the center between the top and bottom layers with ON = 1.77 |e|. The remaining 18 delocalized  $\pi$  bonds are divided into four categories over the  $\sigma$ -skeleton, including 2 5c-2e  $\pi$  bonds at two corners on the left and right with ON = 1.88 |e|, 8 6c-2e  $\pi$  bonds on the top and bottom layers around the central region with ON = 1.82-1.83 |e|, 2 7c-2e  $\pi$  bonds over two  $B_6$  hexagonal pyramids at the center on the top and bottom layers with ON = 1.79 |e|, and 6 17c-2e  $\pi$  bonds at the upper and down ends with ON = 1.80-1.92 |e|. Such a bonding pattern in an overall symmetry of  $D_2$  follows the universal bonding pattern of  $\sigma + \pi$  double delocalization observed in bilayer  $B_{48}$ ,  $B_{54}$ , and  $B_{62}$ <sup>[11]</sup> and the  $B_n^q$  borospherene family ( $n = 36-42$ ,  $q = n-40$ ).<sup>[3,5]</sup>

$D_2 B_{68}$  (III) exhibits a similar bonding pattern (Figure 4b) with  $D_2 B_{66}$  (II). It has 84  $\sigma$ -bonds in total, including 4 interlayer 2c-2e  $\sigma$  bonds between eight inward-buckled B atoms in the top and bottom layers, 28 2c-2e  $\sigma$  bonds on the waist, 48 3c-2e  $\sigma$  bonds on the top and bottom layers, 2 5c-2e  $\sigma$  bonds at two corners on the left and right, and 2 7c-2e  $\sigma$  bonds on the two  $B_6$  hexagonal pyramids on the top and bottom layers at the center. The corresponding  $\pi$  system includes 2 5c-2e  $\pi$  bonds at two corners on the left and right, 8 6c-2e  $\pi$  bonds around the central region on the top and bottom layers, 2 7c-2e  $\pi$  bonds at the centers of the top and bottom layers, and 6 18c-2e  $\pi$  bonds around the upper and down ends of the bilayer structure. The universal  $\sigma + \pi$  double delocalization bonding pattern renders 3D aromaticity to the bilayer  $B_n$  cluster series with  $n = 48, 54, 60, 64, 66, 68, 70$ , and 72, as evidenced by their calculated negative NICS values of NICS =  $-41.5 \sim -5.7$  ppm at their geometrical centers (Table 1). These bilayer species possess the large calculated HOMO-LUMO gaps between  $\Delta E_{gap} = 1.13 \sim 1.98$  eV, showing their high chemical stabilities in gas phases, with  $C_2 B_{54}$  and  $C_1 B_{72}$  possessing the largest  $\Delta E_{gap} = 1.98$  eV and 1.55 eV, respectively (Table 1).

## Spectral Simulations

To facilitate future spectral characterizations of these bilayer species, we simulate the infrared (IR), Raman, and UV-vis spectra of  $D_2 B_{66}$  (II) and  $D_2 B_{68}$  (III) in Figure 5.  $B_{66}$  (II) exhibits five strong IR peaks at 231  $\text{cm}^{-1}$  ( $b_2$ ), 273  $\text{cm}^{-1}$  ( $b_2$ ), 958  $\text{cm}^{-1}$  ( $b_3$ ), 1135  $\text{cm}^{-1}$  ( $b_2$ ), and 1433  $\text{cm}^{-1}$  ( $b_3$ ), while  $D_2 B_{68}$  (III) possesses three intensive IR active modes at 824  $\text{cm}^{-1}$  ( $b_3$ ), 936  $\text{cm}^{-1}$  ( $b_3$ ), and 1435  $\text{cm}^{-1}$  ( $b_3$ ), respectively.  $B_{66}$  (II) has two major Raman active vibrational modes at 811  $\text{cm}^{-1}$  (a) and 1421  $\text{cm}^{-1}$  (a) and  $D_2 B_{68}$  (III) exhibits two strong Raman bands at 839  $\text{cm}^{-1}$  (a) and 851  $\text{cm}^{-1}$  (a), respectively. The weak vibrational modes at 275  $\text{cm}^{-1}$  (a) in  $D_2 B_{66}$  (II) and 255 and 320  $\text{cm}^{-1}$  in  $D_2 B_{68}$  (III)

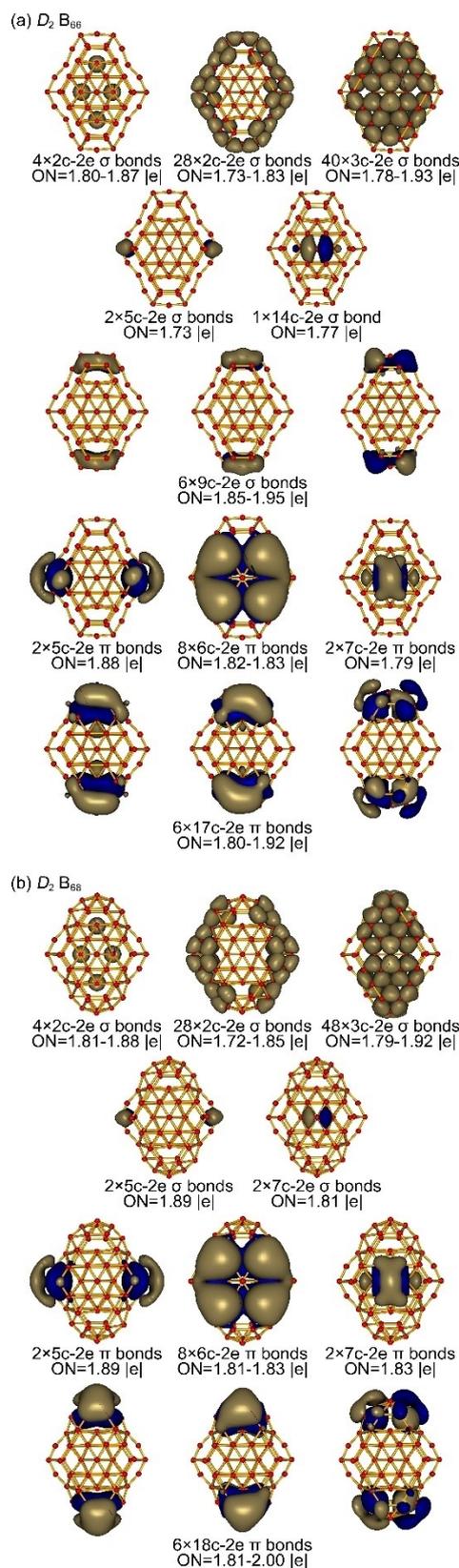


Figure 4. AdNDP bonding patterns of (a)  $D_2 B_{66}$  (II) and (b)  $D_2 B_{68}$  (III).

represent typical “radial breathing modes” (RBMs)<sup>[18]</sup> of the two bilayer structures at PBE0/6-31G(d) which can be used to characterize hollow boron nanostructures.

The strong UV absorption peak of  $D_2 B_{66}$  (II) occurs at 324 nm ( $^1B_3$ ) which mainly originates from the  $S_0 \rightarrow S_{327}$  electronic excitation with major contributions from HOMO-16  $\rightarrow$  LUMO+2 (25%) and HOMO-4  $\rightarrow$  LUMO+11 (19%) transitions (Table S1). The other strong absorptions at 375 ( $^1B_3$ ), 388 ( $^1B_3$ ) and 444 ( $^1B_3$ ) nm also correspond to the excitations from the occupied inner shells to unoccupied frontier orbitals. The strongest UV absorption peak of  $D_2 B_{68}$  (III) at 315 nm ( $^1B_3$ ) corresponds to the electronic transition of  $S_0 \rightarrow S_{365}$  with major contributions from HOMO-9  $\rightarrow$  LUMO+8 (22%) and HOMO-11  $\rightarrow$  LUMO+5 (25%) (Table S1).

## Summary

Based on extensive GM searches and manual structural constructions, we have expanded in this work the bilayer boron cluster series from the previously reported  $D_{2h} B_{48r}$ ,  $C_2 B_{54r}$ ,  $C_{2h} B_{60r}$  and  $C_1 B_{62}^{[11]}$  to  $C_2 B_{64}$  (I),  $D_2 B_{66}$  (II),  $D_2 B_{68}$  (III),  $C_1 B_{70}$  (IV), and  $C_1 B_{72}$  (V) where bilayer configurations dominate and predicted the bilayer to core-shell transition at  $B_{74}$  where a core-shell isomer starts to prevail in thermodynamics. The bilayer I-V all contain an elongated  $B_{46}$  hexagonal prism at the center which features four effective interlayer 2c-2e B-B  $\sigma$  bonds formed between the top and bottom layers. Such 3D aromatic bilayer boron nanoclusters follow the universal bonding pattern of  $\sigma + \pi$  double delocalization and may be used as building blocks to form stable 2D free-standing bilayer borophenes partially passivated via the formation of effective interlayer B-B  $\sigma$  bonds and weak interlayer  $\pi$ - $\pi$  stacking.

## Theoretical procedures

The GMs of  $B_{64r}$ ,  $B_{66r}$ ,  $B_{68r}$ ,  $B_{70r}$  and  $B_{72r}$  were searched using the TGMIn program<sup>[19]</sup> at the PBE level,<sup>[20]</sup> in conjunction with manual structural constructions based on the previously predicted bilayer  $B_{48r}$ ,  $B_{54r}$ ,  $B_{60r}$ , and  $B_{62}^{[9f,11]}$ . Low-lying isomers were firstly reoptimized at the PBE0/6-31G(d), with vibrational frequencies checked (Table 1) and zero-point corrections included. To achieve more reliable relative energies, the five lowest-lying isomers of  $B_{64r}$ ,  $B_{66r}$ ,  $B_{68r}$ ,  $B_{70r}$ , and  $B_{72r}$  were further reoptimized at the PBE0/6-311+G(d)<sup>[13]</sup> and TPSSh/6-311+G(d)<sup>[13a,15]</sup> levels, respectively. Detailed chemical bonding analyses were carried out on the high-symmetry  $D_2 B_{66}$  (II) and  $D_2 B_{68}$  (III) utilizing the adaptive natural density partitioning (AdNDP)<sup>[17]</sup> approach at PBE0/6-31G, with the bonding patterns visualized utilizing the Molekel software.<sup>[21]</sup> The IR and Raman spectra of  $D_2 B_{66}$  (II) and  $D_2 B_{68}$  (III) were simulated at PBE0/6-31G(d) level and their UV-vis absorption spectra calculated using the time-dependent density functional method (TD-DFT-PBE0).<sup>[22]</sup> All the calculations in this work were done using the Gaussian16 package.<sup>[23]</sup> Extensive Born-Oppenheimer molecular dynamics (BOMD) simulations were performed on the  $D_2 B_{66}$  (II) and  $D_2 B_{68}$

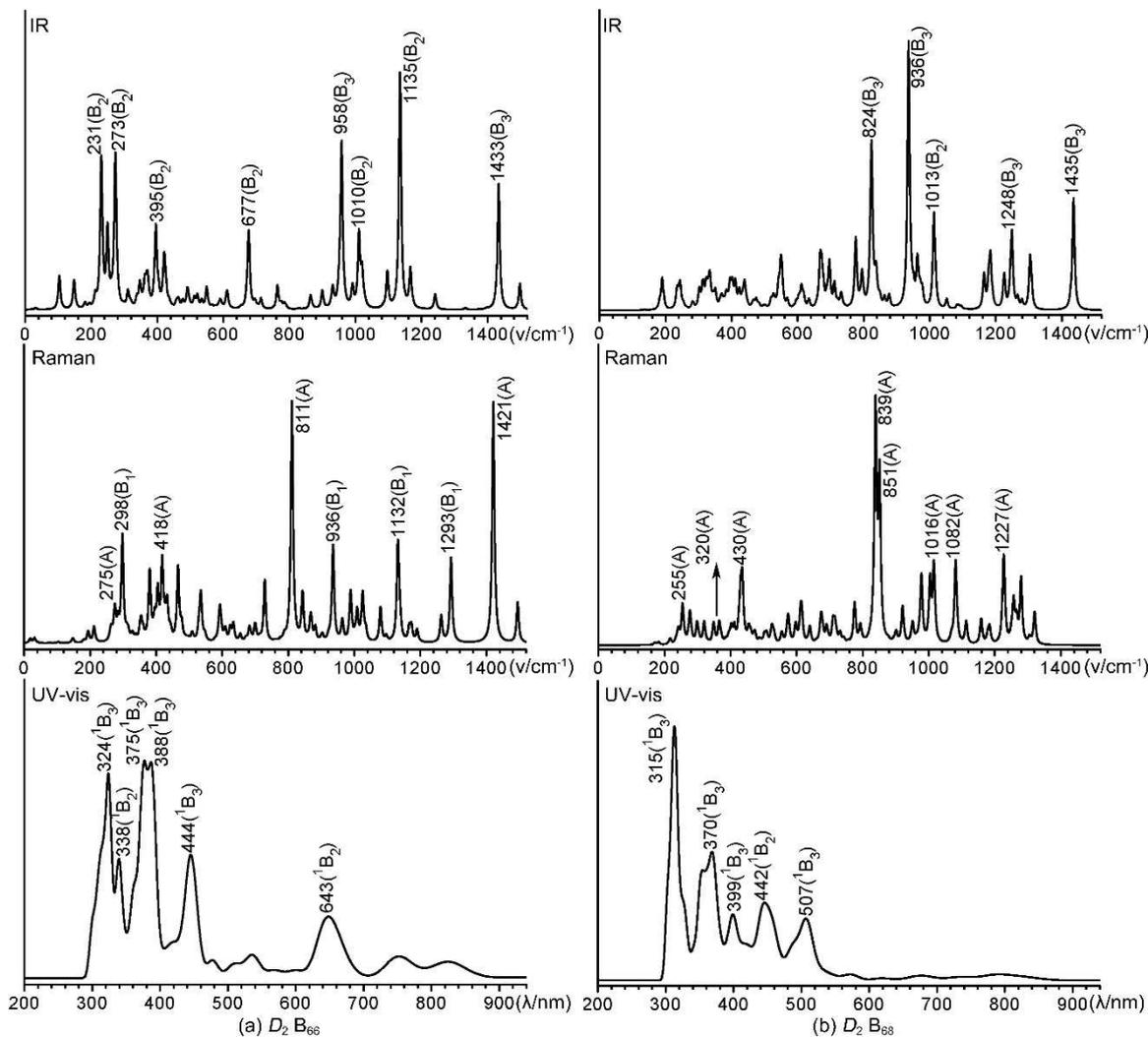


Figure 5. Simulated IR, Raman, and UV-vis of (a)  $D_2 B_{66}$  (II) and (b)  $D_2 B_{68}$  (III) at PBE0/6-31G(d) level.

(III) at 700 K for 30 ps, using the software suite CP2K<sup>[24]</sup> with the GTH-PBE pseudopotential and DZVP-MOLOPT-SR-GTH basis set.

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## Conflict of Interest

The authors declare no conflict of interest.

**Keywords:** Boron Clusters · Density Functional Theory · Structures, Bonding · Three-Dimensional Aromaticity · Bilayer Borophenes

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